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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/565,128	01/17/2006	Tatsuo Shimizu	06.22.01.P	7793
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EXAMINER				
BARROW, AMANDA J				
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Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Office Action Summary

Application No.

10/565,128

Applicant(s)

SHIMIZU ET AL.

Examiner

AMANDA BARROW

Art Unit

1795

Period for Reply -- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

- 1) ☒ Responsive to communication(s) filed on 16 September 2010.
- 2a) ☒ This action is **FINAL**. 2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

- 4) ☒ Claim(s) 1-10 is/are pending in the application.
- 4a) Of the above claim(s) 2, 3, 7 and 9 is/are withdrawn from consideration.
- 5) ☐ Claim(s) _____ is/are allowed.
- 6) ☒ Claim(s) 1, 4-6, 8 and 10 is/are rejected.
- 7) ☐ Claim(s) _____ is/are objected to.
- 8) ☐ Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on _____ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

Priority under 35 U.S.C. § 119

- 12) ☒ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☒ All b) ☐ Some * c) ☐ None of:
1. ☒ Certified copies of the priority documents have been received.
 2. ☐ Certified copies of the priority documents have been received in Application No. _____.
 3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

* See the attached detailed Office action for a list of the certified copies not received.

Attachment(s)

- 1) ☐ Notice of References Cited (PTO-892)
- 2) ☐ Notice of Draftperson's Patent Drawing Review (PTO-948)
- 3) ☒ Information Disclosure Statement(s) (PTO/SB-08)
Paper No(s)/Mail Date 10/15/2010
- 4) ☐ Interview Summary (PTO-413)
Paper No(s)/Mail Date _____
- 5) ☐ Notice of Informal Patent Application
- 6) ☐ Other: _____

DETAILED ACTION

Status of Application

1. The Applicant's amendment filed on 9/16/2010 was received. Claim 10 was amended. Claims 1, 4-6, 8 and 10 are pending. Claims 2, 3, 7 and 9 have been previously withdrawn.
2. The texts of those sections of Title 35, U.S.C. code not included in this action can be found in the prior Office Action issued on 9/4/2009.

Claim Rejections - 35 USC § 112

3. The rejection on claim 10 under 35 U.S.C. 112, first paragraph, as failing to comply with the written description requirement is withdrawn because the claim has been amended.
4. The rejection under claim 1 under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention is withdrawn because the Applicants' arguments are persuasive.
5. The rejection on claim 8 under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention is maintained.

Claim 8 recites that the electrode active material has a density less than or equal to 1.4 grams per cubic centimeter in an upper region. It is not clear to the Examiner where the "upper region" is located. Appropriate correction is required.

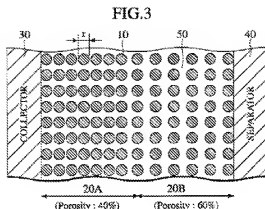
Furthermore, it is unclear whether the density referred to in the claim refers to the density of the actual material or the overall density (packing density) of all of the carbon material. The specification notes that the density is the “void percentage” and thus it is clear that what is meant is the packing density/void percentage/porosity. Appropriate correction is required.

Claim Rejections - 35 USC § 103

6. The claim rejections under 35 U.S.C. 103(a) as being unpatentable over Tajima et al. JP 40-1091167A in view of Tanjo (US 2002/0028380) on claims 1, 4 and 6 are maintained.

Regarding claim 1, Tajima discloses a carbon electrode (“current collecting structure”) that includes an electroconductive substrate (“current collecting substrate”) that has a carbon material directly deposited on the conductive substrate (“current collecting substrate”) without the use of binders (see abstract; column 1, lines 15-50).

Tajima does not disclose the density (void percentage) of the carbon material on the current collecting substrate; however, Tanjo discloses analogous art of a battery having a collector 30 (“current collector substrate”) with an active material layer 20 containing the positive electrode active material (paragraph 44). Tanjo teaches that the active material layer 20 may have a plurality of active material layers having different porosities (paragraph 51). The electrode may have a two layer structure in which the layer closer to the separator 20B (“the upper region”) has a porosity larger than the active material layer 20A closer to the collector. In other words, the active material has a higher density/lower void percentage in the area near the current collector 30 than in the area closer to the separator (“upper region”). This is illustrated below in Figure 3:



Tanjo teaches that the amount of the electrolytic solution 50 in the vicinity of the separator 40 can be increased and the migration power of lithium ion can be increased by making the porosity of the active material layer 20B adjacent to the separator 40 large. Moreover, a usage rate of the active material 10 in the vicinity of the collector 30 can be increased by making the porosity in the vicinity of the collector 30 small. Thus, the power density can be effectively increased by balancing the diffusion in the positive electrode active material 10 and the migration in the electrolytic solution 50. Since the energy density is influenced by an average porosity and the active material amount of the active material layer 20, the power density can be increased without sacrificing the energy density by appropriately adjusting the average porosity and the amount of the active material (paragraph 51).

Therefore, it would have been obvious to a person of ordinary skill in the art to modify the carbon layer of Tajima to have a higher density/lower void percentage near the collector as compared to the "upper region" near the separator because Tanjo teaches such a configuration (Figure 3) and that this allows for an increase in migration power of the lithium ion and an increased usage rate of the active material in the vicinity of the collector which effectively increases the power density (paragraph 51).

Regarding claim 4, Tajima discloses that a charge carrier capable of reversible intercalation and deintercalation ("electrode active material") is supported on the carbon material deposited on the substrate (see abstract).

Regarding claim 6, Tajima discloses that the electrode is to be used in a battery (see abstract).

7. The rejections under 35 U.S.C. 103(a) as being unpatentable over Tajima et al JP 40-1091167A in view of Tanjo (US Patent Application 2002/0028380) on claims 8 and 10 are maintained.

Regarding claim 8, Tajima discloses a carbon electrode that includes an electroconductive substrate ("current collecting substrate") that has a carbon material ("electrode active material") directly deposited on the electroconductive substrate ("current collecting substrate") without the use of binders (see abstract).

Tajima does not disclose the density (void percentage) of the carbon material on the current collecting substrate; however, Tanjo discloses analogous art of a battery with a collector 30 and an active material layer 20 containing the positive electrode active material (paragraph 44). Tanjo teaches that the active material layer 20 may have a plurality of active material layers having different porosities (paragraph 51). The electrode may have a two layer structure in which the layer closer to the separator 20B ("the upper region") has a porosity larger than the active material layer 20A closer to the collector. In other words, the active material has a higher density/lower void percentage in the area near the current collector 30 than in the area closer to the separator ("upper region").

Tanjo teaches that the amount of the electrolytic solution 50 in the vicinity of the separator 40 can be increased and the migration power of lithium ion can be increased by making the porosity of the active material layer 20B adjacent to the separator 40 ("upper region") large. Moreover, a usage rate of the active material 10 in the vicinity of the collector 30 can be increased by making the porosity in the vicinity of the collector 30 small. Thus, the power density can be effectively increased by balancing the diffusion in the positive electrode active material 10 and the migration in the electrolytic solution 50. Since the energy density is influenced by an average porosity and the active material amount of the active material layer 20, the power density can be increased without sacrificing the energy density by appropriately adjusting the average porosity and the amount of the active material (paragraph 51).

Therefore, it would have been obvious to a person of ordinary skill in the art to optimize the density/void percentage of the active material layer of Tajima to have a specific density/ void percentage in active material layer 20A ("upper region") near the separator because Tanjo teaches such a configuration (Figure 3) and that this allows for an increase in migration power of the lithium ion and an increased usage rate of the active material in the vicinity of the collector which effectively increases the power density (paragraph 51). Furthermore, since the energy density is influenced by an average porosity and the active material amount of the active material layer 20, the power density can be increased without sacrificing the energy density by appropriately adjusting the average porosity and the amount of the active material (paragraph 51 - Tanjo). The discovery of an optimum value of a known result effective variable, without producing any new or unexpected results, is within the ambit of a person of ordinary skill in the art. See *In re Boesch*, 205 USPQ 215 (CCPA 1980) (see MPEP § 2144.05, II.).

Regarding claim 10, Tajima discloses that a charge carrier capable of reversible intercalation and deintercalation ("conductive material") is supported on the carbon material deposited on the substrate (abstract).

8. The claim rejection under 35 U.S.C. 103(a) as being unpatentable over Tajima et al. in view of Tanjo et al. as applied to claims 1, 4 and 6 above and further in view of Nakai et al. (US Patent Application 2002/0122983 A1) on claim 5 is maintained.

Regarding claim 5, Tajima does not disclose the mean particle diameter of the material capable of being reversibly intercalated and deintercalated ("electrode active material"); however, Nakai teaches that in using electrode active materials that have a small average particle diameter (0.1 to 2 microns), the reaction area of the active material is optimized therefore improving the power characteristic without enlarging the size of the battery (paragraph 13).

Therefore, it would have been obvious to a person of ordinary skill in the art to optimize the particle diameter of the electrode active material of Tajima because Nakai teaches that in using particles with small diameters (0.1 to 2 microns), the reactive area of the active material is optimized therefore improving the power characteristic without enlarging the size of the battery (paragraph 13). The discovery of an optimum value of a known result effective variable, without producing any new or unexpected results, is within the ambit of a person of ordinary skill in the art. See *In re Boesch*, 205 USPQ 215 (CCPA 1980) (see MPEP § 2144.05, II.).

Response to Arguments

9. Applicant's arguments filed 9/16/2010 have been fully considered but they are not persuasive.

Applicant's remaining principal arguments are

(a) Applicant respectfully submits that the phrase "upper region" in claim 8 is not indefinite in view of paragraph 27 of the present Application.

(b) The use of the term "density" in claim 8 is not indefinite as it is clear within the specification (see paragraphs 26 and 27).

(c) To be motivated to combine the references, one skilled in the art would have to look at Tajima and Tanjo and determine that some benefit would result from using carbon on a current collecting substrate and having the carbon at two different densities. The Applicant submits that no such motivation exists within the art.

(d) Tanjo does not disclose the use of carbon as the electrode active material as claimed in Applicant's claim 1. The failure of Tanjo to disclose or even consider the use of carbon overcomes the rejection.

(e) Tanjo discloses the use of an active material combined with a binder which would result in an electrode structure that utilizes a binder contrary to the limitations of claim 1.

In response to Applicant's arguments, please consider the following comments.

(a) Limitations appearing in the specification but not recited in the claim are not read into the claim. See *In re Zletz*, 893F.2d 319, 321-22, 13 USPQ2d, 1320, 1322 (Fed. Cir. 1989).

The Examiner withdrew the 112 rejection on claim 1 because the Applicant's argument regarding the differentiation between the upper and lower regions was persuasive (the "higher density" is described in claim 1 as being "near the current collecting substrate" and the "lower density" is described as being "in the upper region," thus differentiating between the area). No such limitations appear in claim 8; as such, the 112 rejection on claim 8 is maintained.

(b) Limitations appearing in the specification but not recited in the claim are not read into the claim. See *In re Zletz*, 893F.2d 319, 321-22, 13 USPQ2d, 1320, 1322 (Fed. Cir. 1989). Claim 8 recites that the electrode active material has a density less than or equal to 1.4 grams per cubic centimeter in an upper region. Without reading the specification, it is unclear whether the density referred to in the claim refers to the density of the actual material or the overall density (packing density) of all of the carbon material.

(c) Tanjo provides the motivation to modify the carbon later of Tajima. Tanjo discloses an active material layer 20 of an electrode may have a plurality of active material layers having different porosities (paragraph 51). The electrode may have a two layer structure in which the layer closer to the separator 20B has a porosity larger than the active material layer 20A closer to the collector and illustrates this configuration in Figure 3. Tanjo teaches that *the amount of the electrolytic solution 50 in the vicinity of the separator 40 can be increased and the migration power of lithium ion can be increased by making the porosity of the active material layer 20B adjacent to the separator 40 large. Moreover, a usage rate of the active material 10 in the vicinity of the collector 30 can be increased by making the porosity in the vicinity of the collector 30 small. Thus, the power density can be effectively increased by balancing the diffusion in the positive electrode active material 10 and the migration in the electrolytic solution 50. Since the*

energy density is influenced by an average porosity and the active material amount of the active material layer 20, the power density can be increased without sacrificing the energy density by appropriately adjusting the average porosity and the amount of the active material (paragraph 51). Thus, it is evident that Tanjo provides multiple motivational reasons to modify the density distribution of the electrode active material layer of Tajima.

(d) The rejection made was not a 102 rejection; therefore, the reference may not be attacked by itself: one cannot show nonobviousness by attacking references individually where the rejections are based on combinations of references. In re Keller, 642 F.2d 413, 208 USPQ 871 (CCPA 1981); In re Merck & Co., Inc., 800 F.2d 1091, 231 USPQ 375 (Fed. Circ. 1986). Furthermore, the Examiner did not rely upon Tanjo to disclose the use of carbon; Tanjo was used to modify the density distribution of the active material. Tajima, the primary reference, was used to disclose an electrode with a current collecting substrate having a carbon material formed on the electrode without the use of binders (Tajima: abstract).

(e) The Tanjo reference was not relied upon to disclose the carbon material formed on the electrode without the use of binder; Tajima, the primary reference, was used to do this: when a binding agent is required in addition to the electrode active material, this results in a decrease in the amount of the active material which induces a decrease in energy density (column 1, lines 27-31); thus the carbon electrode of the invention overcomes this by directly depositing a carbon material on an electroconductive electrode substrate via a chemical vapor depositions method (column 1, lines 38-50).

Conclusion

10. All claims are drawn to the same invention claimed in the application prior to the entry of the submission under 37 CFR 1.114 and could have been finally rejected on the grounds and art of record in the next Office action if they had been entered in the application prior to entry under 37 CFR 1.114. Accordingly, **THIS ACTION IS MADE FINAL** even though it is a first action after the filing of a request for continued examination and the submission under 37 CFR 1.114. See MPEP § 706.07(b). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to AMANDA BARROW whose telephone number is (571)270-7867. The examiner can normally be reached on 7:30am-5pm EST. Monday-Friday, alternate Fridays off.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Ula Ruddock can be reached on 571-272-1481. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/AMANDA BARROW/
Examiner, Art Unit 1795

/Ula C Ruddock/
Supervisory Patent Examiner, Art Unit 1795